

American Association of Petroleum Geologists
Visiting Geologists Program

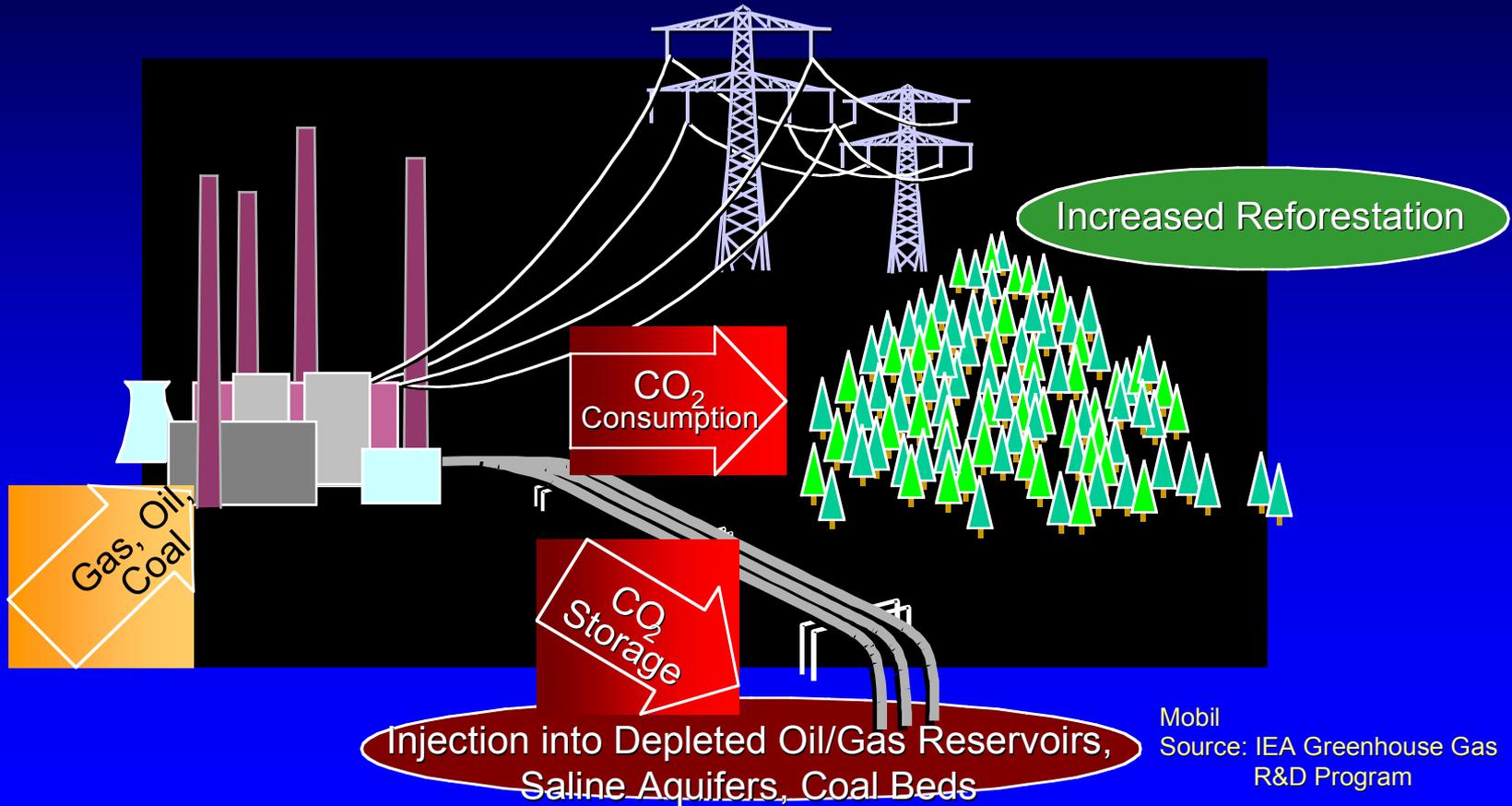
Geologic CO₂ Sequestration



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CO₂ Capture and Storage

Potential for Reducing CO₂ Emissions from Fossil Fuel Power Generation



Mobil
Source: IEA Greenhouse Gas
R&D Program

Why Sequester CO₂?

- Like it or not, fossil fuels will remain the mainstay of energy production well into the 21st century.
- Availability of these fuels to provide clean, affordable energy is essential for the prosperity and security of the United States.
- However, increased concentrations of carbon dioxide (CO₂) due to carbon emissions are expected unless energy systems reduce the carbon emissions to the atmosphere.
- Anthropogenic Green House Gases (GHG's) may be contributing to Global Climate Change.
- To a degree, CO₂ is a useful byproduct. So, why not capture it?

Is The Air Getting Cleaner Or Dirtier?

According to the U.S. Environmental Protection Agency's (EPA) latest Ten-Year Air Quality and Emissions Trends report, there have been significant reductions in all 6 criteria pollutants and reductions are expected to continue.

The pollution reductions between 1986 and 1995 were:

Carbon Monoxide (CO)	down 37%
Lead	down 78%
Nitrogen Dioxide (NO₂)	down 14%
Ozone	down 6%
Particulate Matter (PM-10) . . .	down 22%
Sulfur Dioxide	down 37%

What is Sequestration?

- Capturing and securely storing carbon emitted from the global energy system.

Types of Sequestration?

Ocean Sequestration: Carbon stored in oceans through direct injection or fertilization.

Geologic Sequestration: Natural pore space in geologic formations serve as reservoirs for long term carbon dioxide storage.

Terrestrial Sequestration: A large amount of carbon is stored in soils and vegetation, our natural carbon sinks. Increasing carbon fixation through photosynthesis, slowing down or reducing decomposition of organic matter, and changing land use practices can enhance carbon uptake in these natural sinks.

Geologic Sequestration

Trapping Mechanisms

- **Hydrodynamic Trapping:** carbon dioxide can be trapped as a gas under low-permeability cap rock (much like natural gas is stored in gas reservoirs).
- **Solubility trapping:** carbon dioxide can be dissolved into a liquid – water and/or oil.
- **Mineral Carbonation:** carbon dioxide can react with the minerals, fluids, and organic matter in the geologic formation to form stable compounds/minerals; largely calcium, iron, and magnesium carbonates.

Primary Geologic Sequestration Target Reservoirs

- **Oil and Gas Pools/Fields**
- **Coal Beds**
- **Deep Saline Aquifers**
- **Unconventional Reservoirs – tight gas sands; organic shales; salt domes, etc.**

Long-term storage of CO₂ in underground geologic formations has the potential to be viable in the near-term. Many power plants and other large point sources of CO₂ emissions are located near geologic formations that are amenable to CO₂ storage. Further, in many cases injection of CO₂ into a geologic formation can enhance the recovery of oil and gas which can offset the cost of CO₂ capture.

The use of CO₂ to enhance oil and gas recovery is a common industrial practice. In the year 2000 in the United States, 34 million tons of CO₂ were injected underground as a part of enhanced oil recovery (EOR) and coal bed methane recovery (E-CBM) operations. This is approximately equivalent to the CO₂ emissions from 6 million cars in one year. Research and development in this area will move the technology forward to make it applicable to a wider range of formations.

A novel process which currently experiences a broad interest is the injection of CO₂ in unmineable coalbeds, thus releasing the trapped methane. This process is called Enhanced Gas Recovery (EGR) or Enhanced CoalBed Methane production (ECBM), and is similar to the popular practice of using CO₂ injection to enhance production from oil reservoirs.

With EGR, the injected CO₂ is adsorbed by the coal and stored in the pore matrix of the coal seams, releasing the trapped methane that can be sold for profit. Future work in the area can lead to the design of efficient null-greenhouse-gas-emmission power plants that are fuelled either by mineable coal or by the methane released from the deep coal reservoirs. In this closed CO₂ process, the waste CO₂ produced from the coal or methane-powered plants is injected into the CBM reservoirs to produce more methane, and the cycle continuous.

In addition, a geological sink is established in the coalbeds, virtually eliminating any release of CO₂ to the atmosphere.

Saline formations do not contain oil and gas resources and thus do not offer the value-added benefit of enhanced hydrocarbon production. However, the potential CO₂ storage capacity of domestic saline formations is huge; estimates are on the order of several hundred years of CO₂ emissions.

The primary goal of research in this area is to understand the behavior of CO₂ when stored in geologic formations so that CO₂ can be stored in a manner that is secure and environmentally acceptable.

CO₂ Separation and Capture – The Achilles Heel?

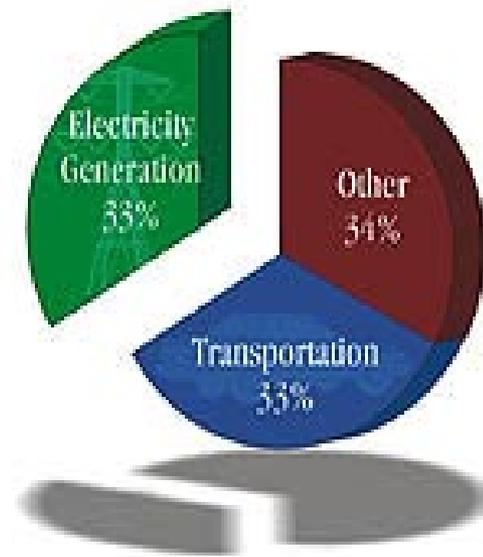
CO₂ is currently recovered from combustion exhaust streams for use as a commodity chemical. However, the cost of CO₂ capture using current technology is much too high (\$100-300/ton) for carbon emissions reduction applications. Research to reduce the cost is in the early stages, and the program is exploring a wide range of technologies, including membranes, solid sorbents, CO₂ capture via the formation of CO₂/water hydrates, and advanced gas/liquid contactors.

Another approach to CO₂ capture is to develop advanced fossil fuel energy conversion processes that exhaust CO₂ in a more concentrated form, significantly reducing the capital and energy penalty cost for CO₂ capture. Efforts in this area being pursued by the program are closely coordinated with DOE's Vision 21 Program.

What are the major sources of CO₂?

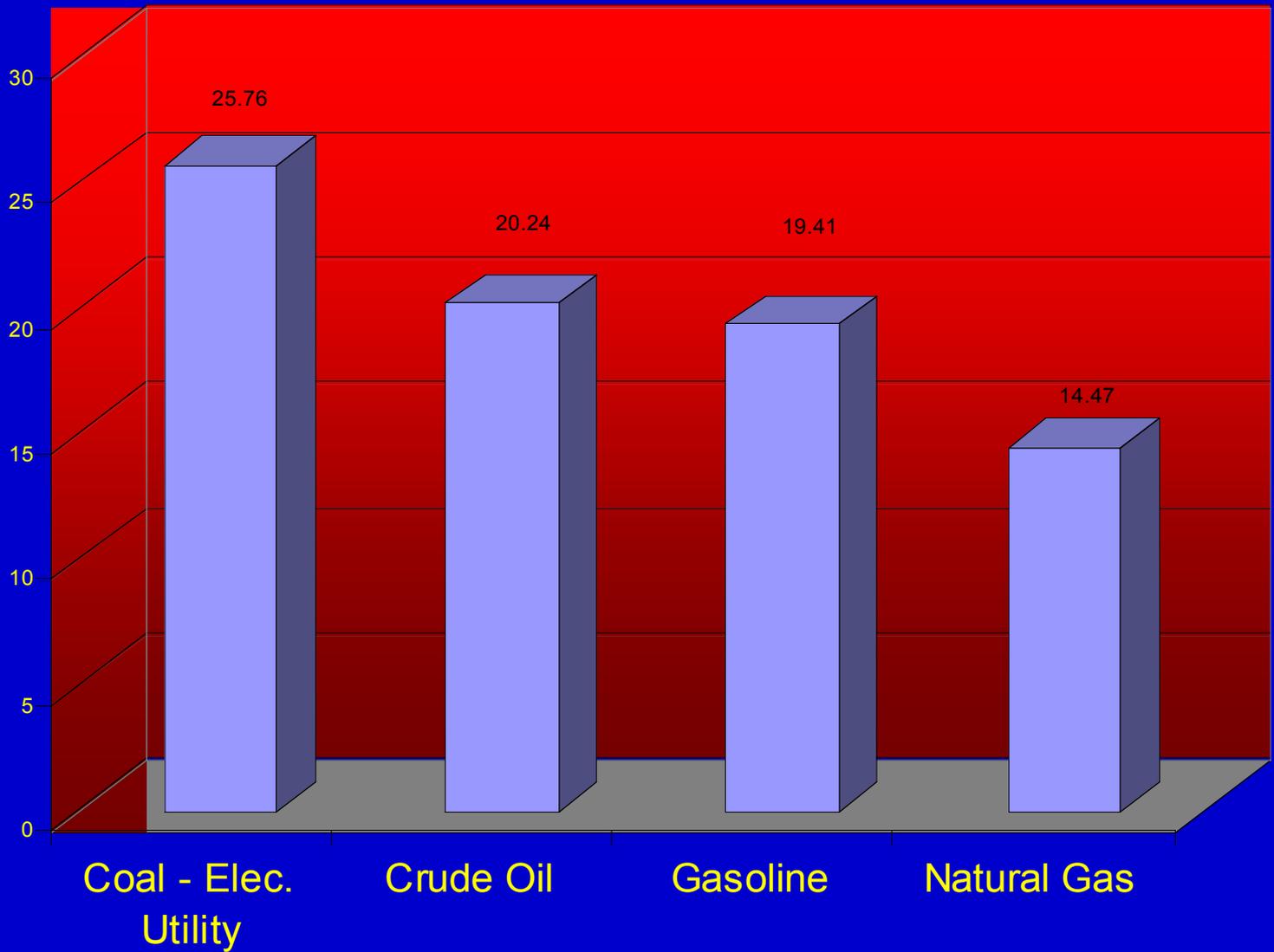
Roughly one third of the United States' carbon emissions come from power plants. These sources would be convenient for CO₂ capture except that most use air-fired combustors, a process that exhausts CO₂ diluted with nitrogen. Flue gas from coal-fired power plants contains 10-12% CO₂ by volume, and flue gas from natural gas combined cycle plants contains from 3-6% CO₂. Concentrated CO₂ (greater than 90%) is needed for most storage, conversion and reuse.

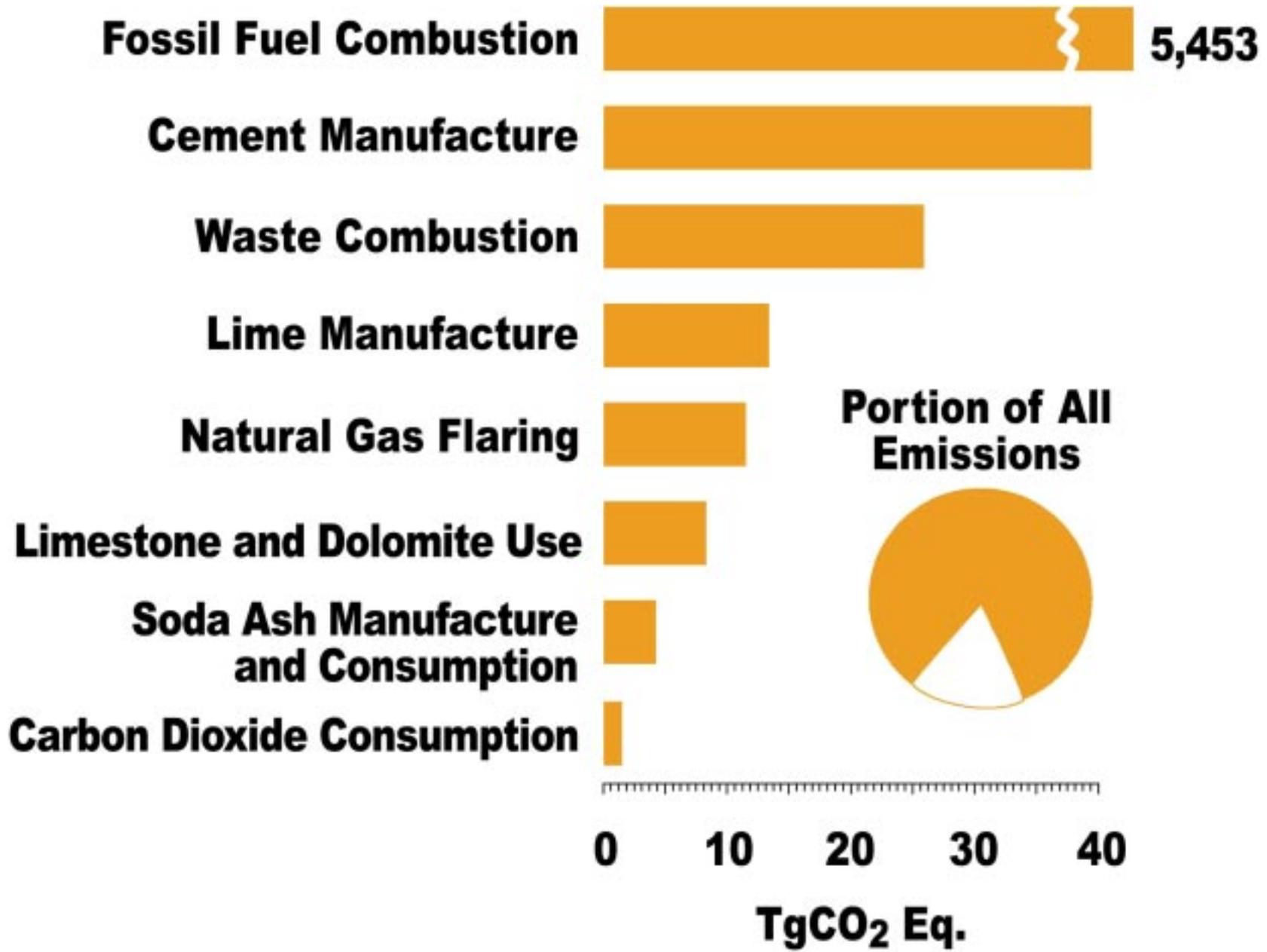
U.S. Carbon Emission Sources

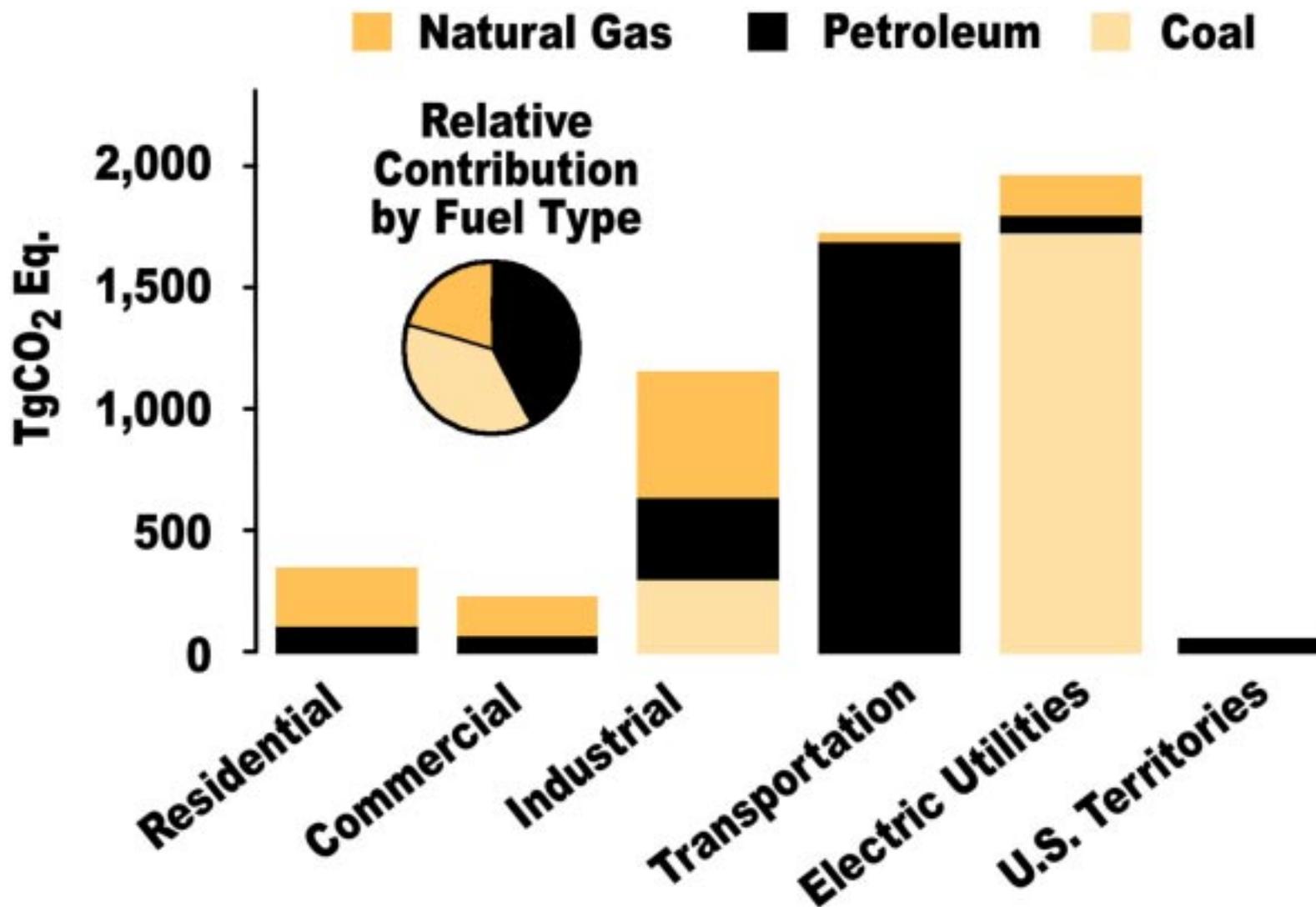


Full Combustion Carbon Coefficients

Tons
Carbon per
Quadrillion
BTU

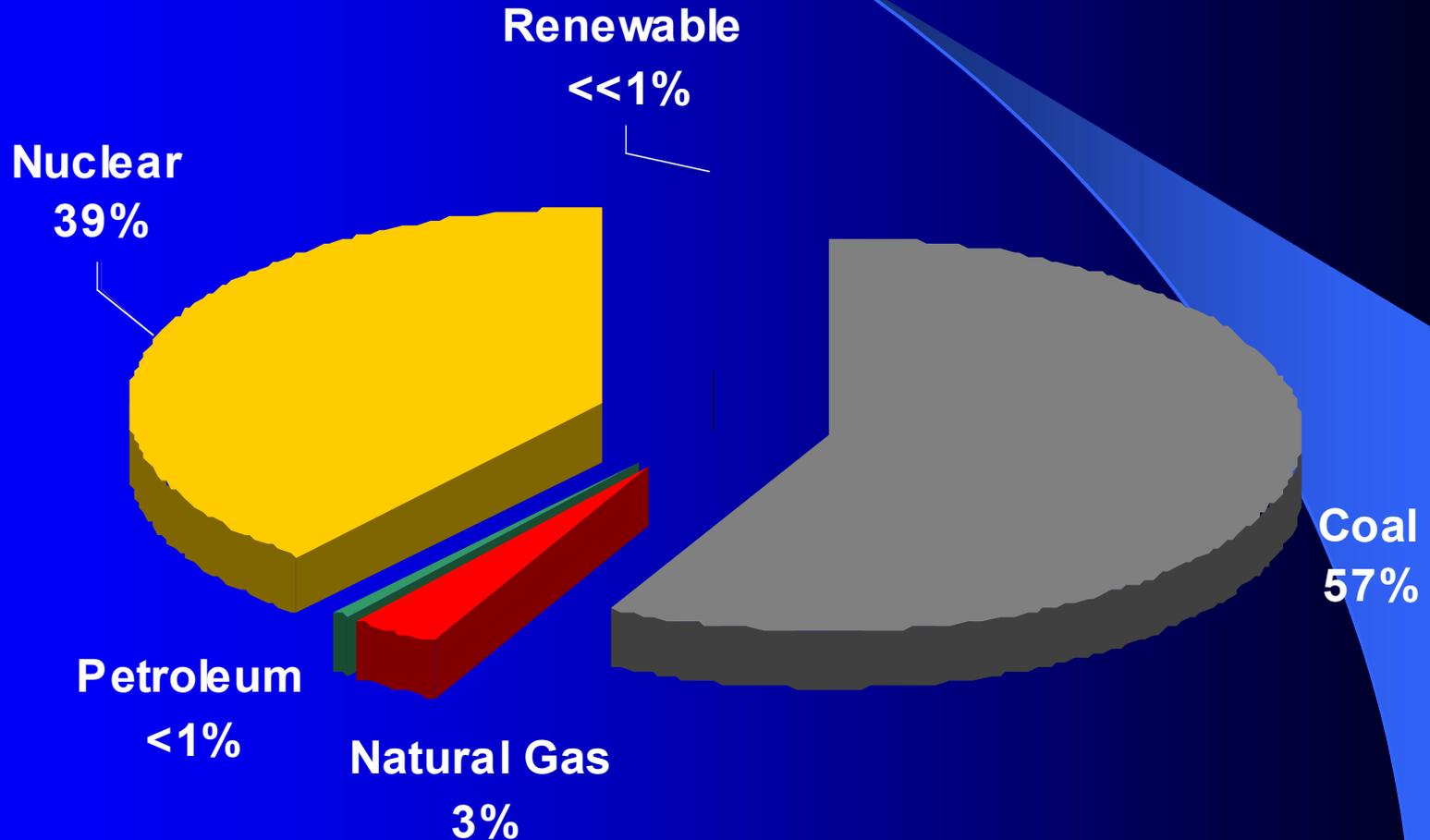






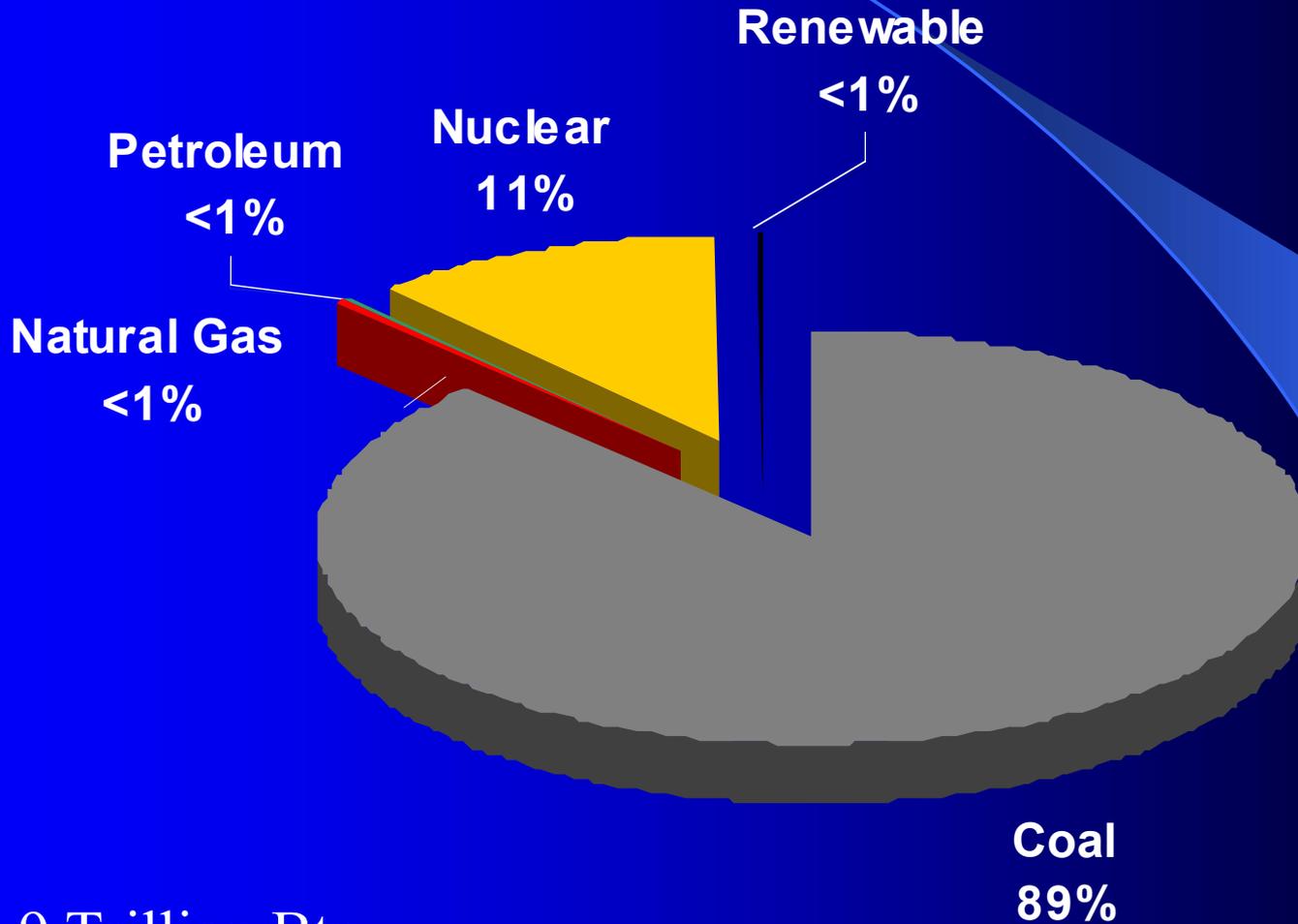
Note: Utilities also include emissions of 0.04 TgCO₂ Eq. from geothermal based electricity generation.

Illinois Electrical Energy Consumption



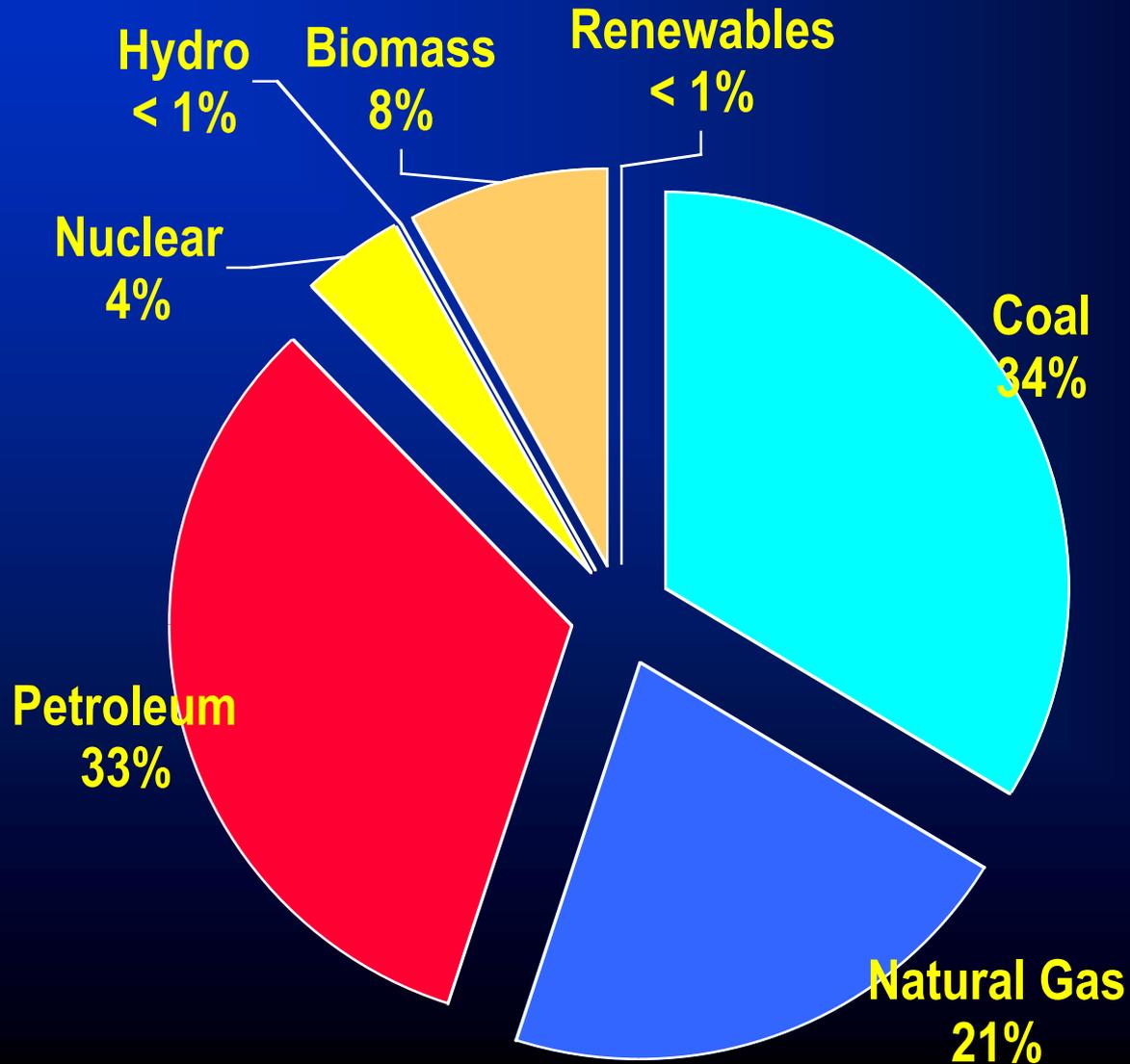
1397.6 Trillion Btu

Ohio Electrical Energy Consumption



1432.9 Trillion Btu

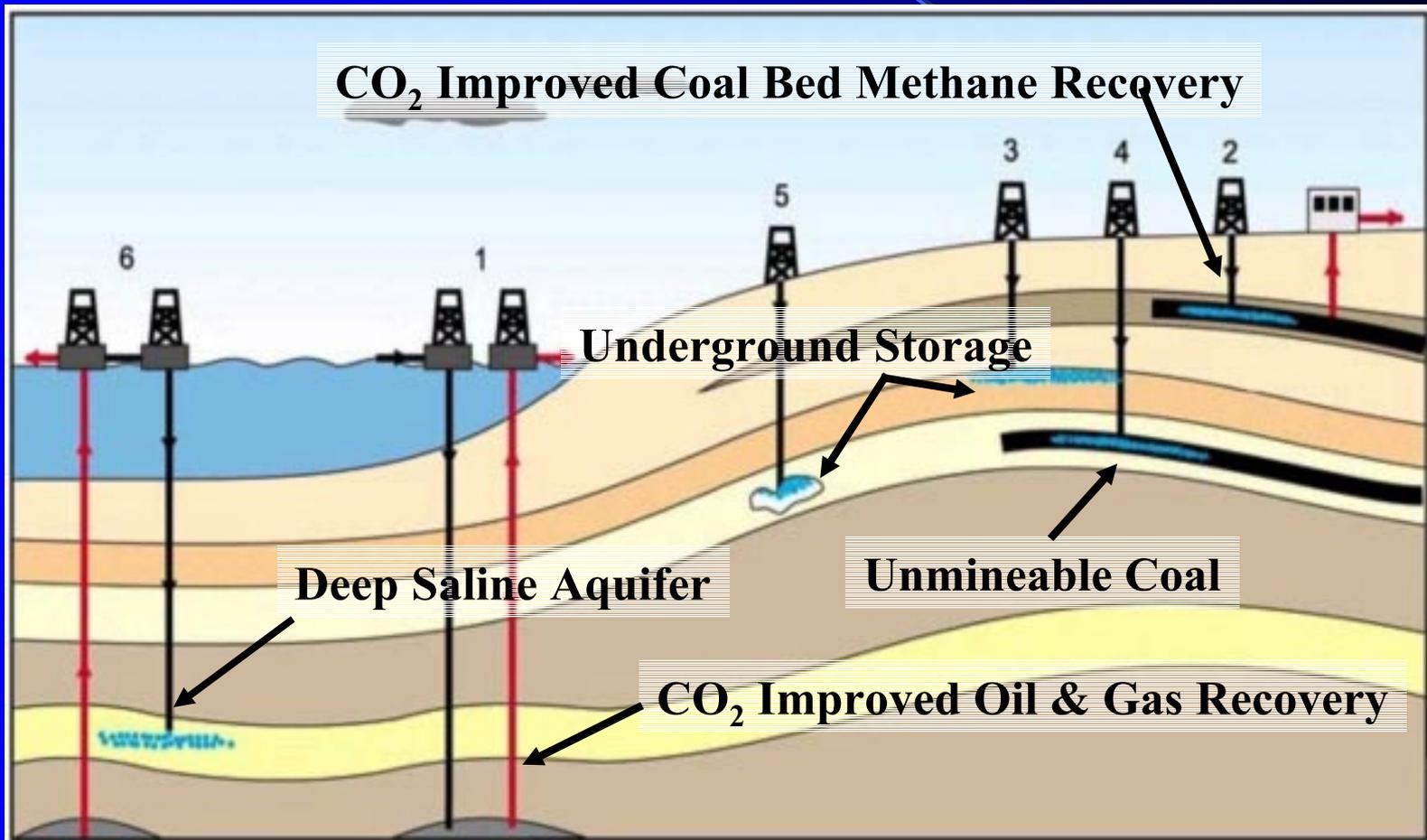
Total Ohio Energy Consumption by Source



CO₂ Sequestration - Sink Characterization

- **Oil Reservoirs**
 - **CO₂ Miscible and Immiscible Flooding**
 - **Reservoir Fluid and Rock Properties**
 - **Geologic and Engineering Data**
- **Coalbeds - Enhanced Methane Recovery**
- **Saline Aquifers**
- **Conventional and Unconventional Gas Reservoirs - Enhanced Gas Recovery?**

CO₂ Geologic Sequestration Options

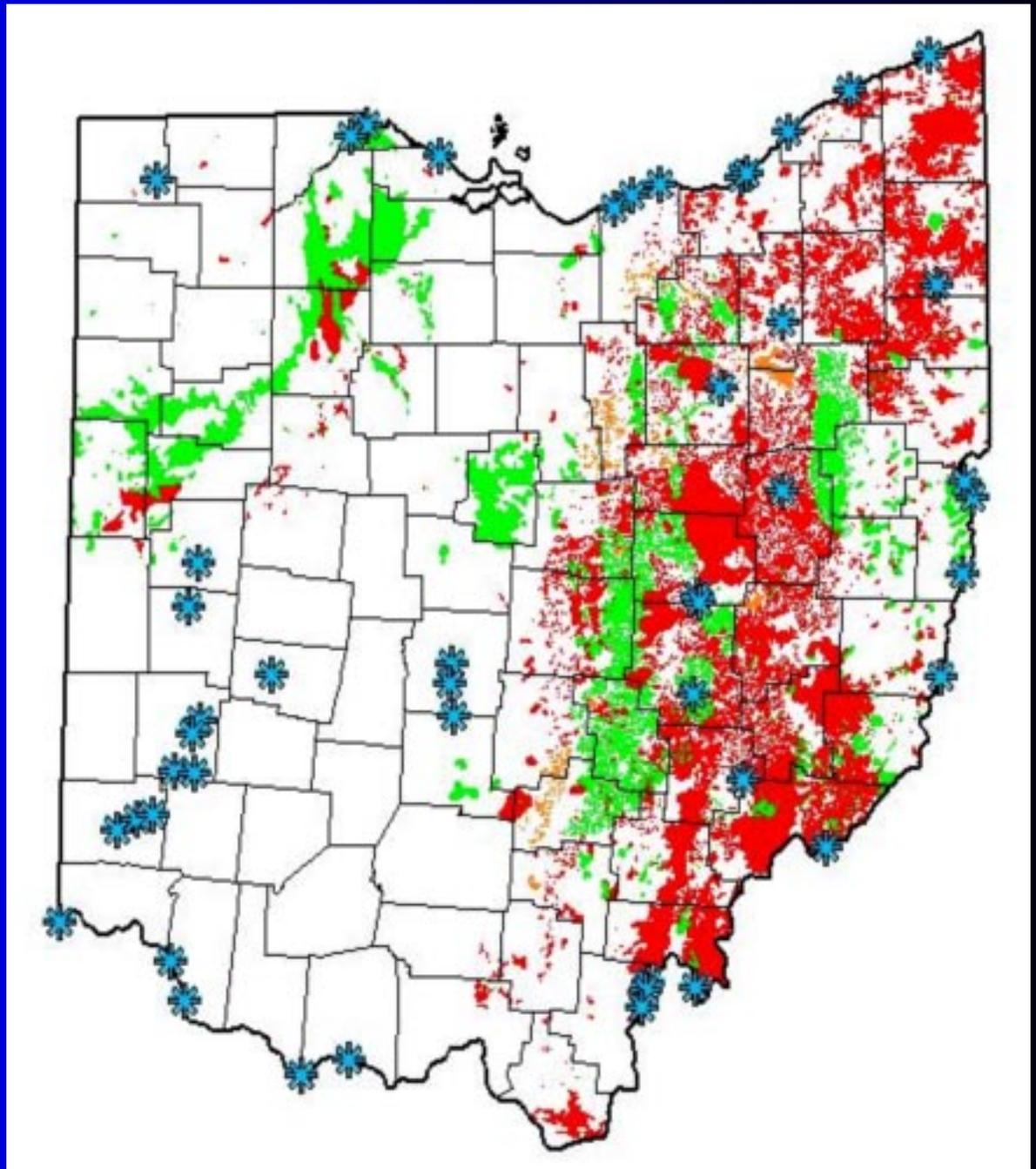


Ohio

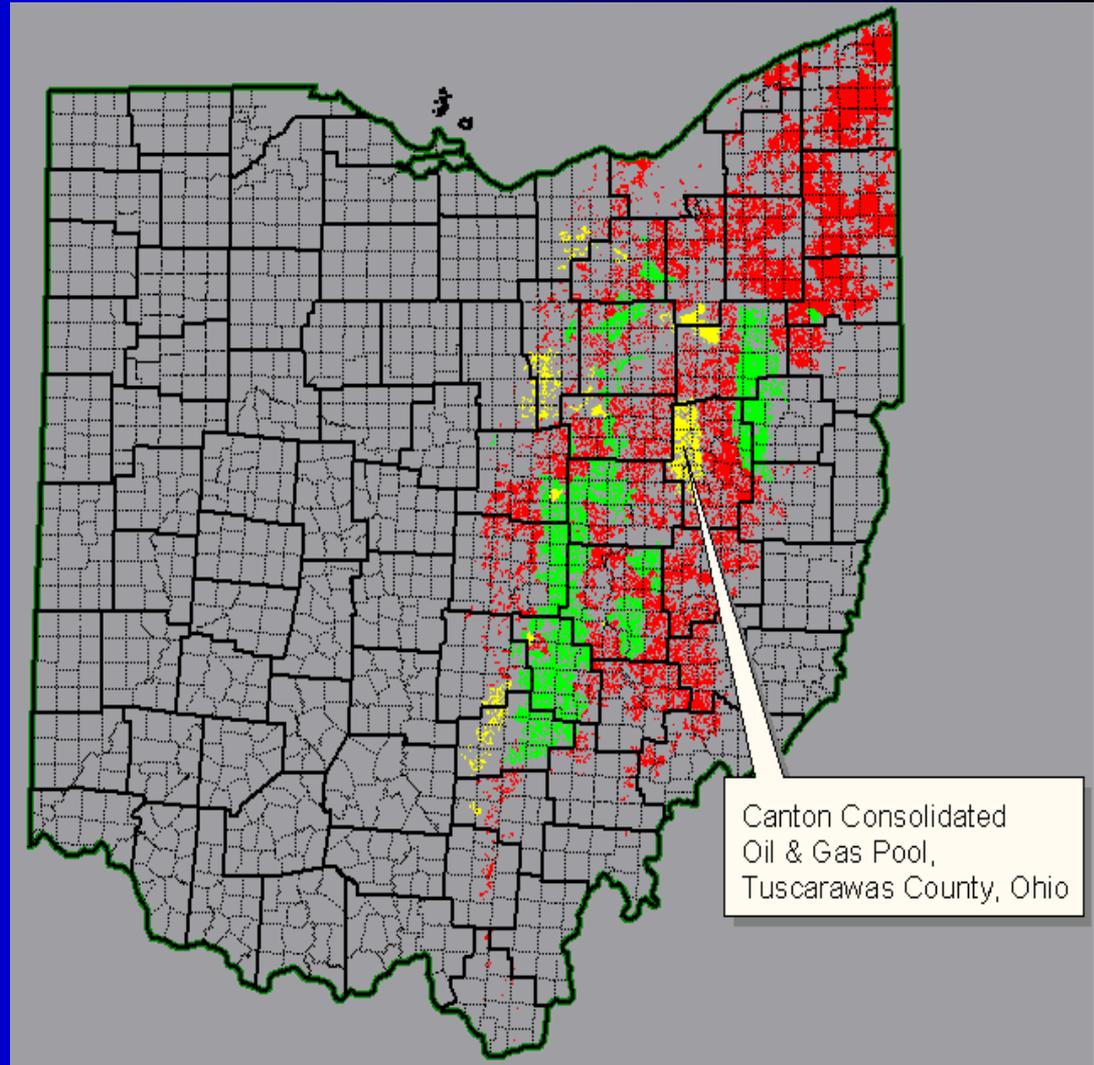
Oil & Gas Fields

And

Power Generating
Plants



The amount of CO₂ sequestration in oil & gas fields can be calculated using geographic information systems (GIS) technology. In this figure, the Clinton sandstone oil & gas pools GIS layer is displayed. Each pool in the GIS layer is represented by a color filled polygon and each of the polygons is tied to a record in the attribute table. Each pool has many different attributes associated with it, such as Average Thickness, Average Porosity, and Original Oil In Place. Using the attributes associated with each polygon, calculations can be made as to how much CO₂ can be sequestered in each oil & gas pool. These calculations are now an attribute associated with each polygon in the GIS. Highlighted in yellow, this pool of the Canton Consolidated oil & gas field can sequester over 51 billion tons of CO₂.





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New Open Print

3D Wells

- Demo - Ashtabula South C
- Demo 1 - Ohio Oil & Gas F
- Demo 2 - Oil & Field Attrib.
- Demo 3 - Assigning Field P
- Demo 4 - Knox Field Upde
- Demo 5 - CO2 Sequestrati

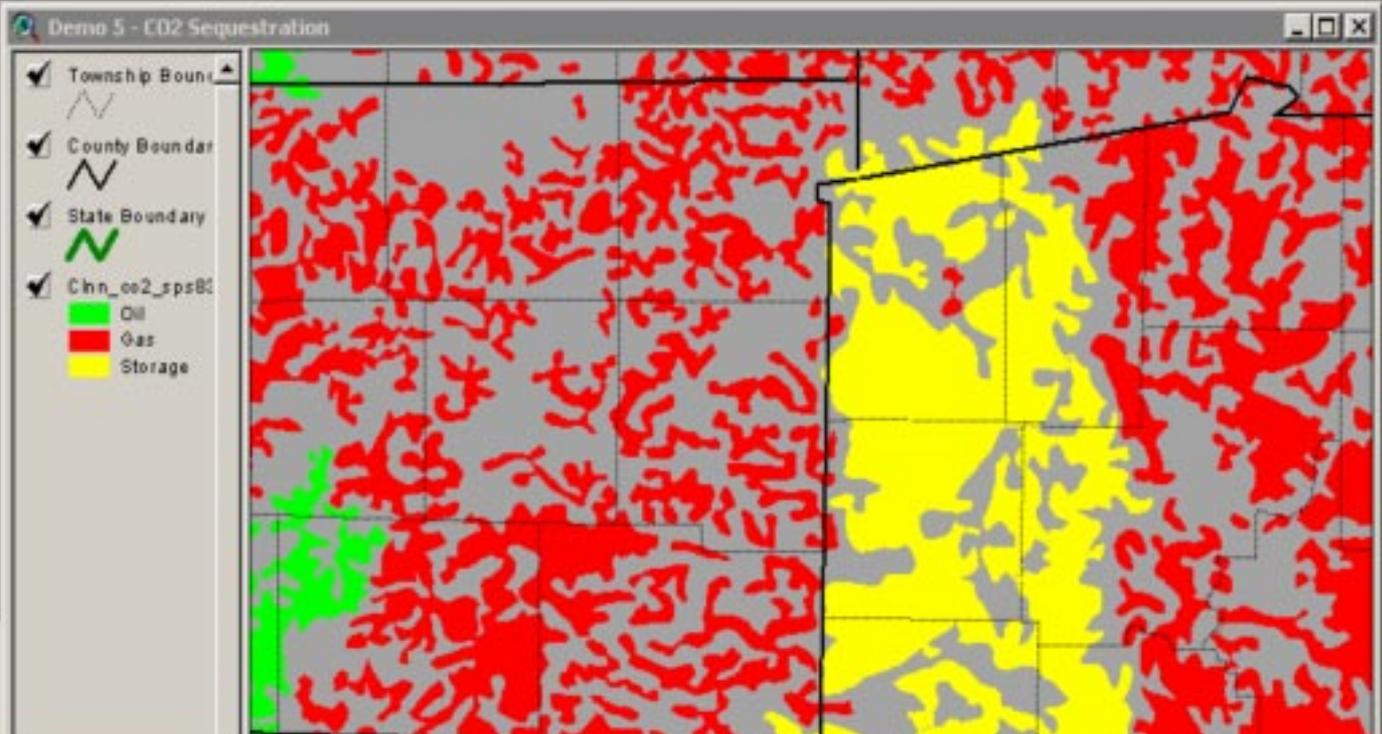
Views

Tables

Charts

Layouts

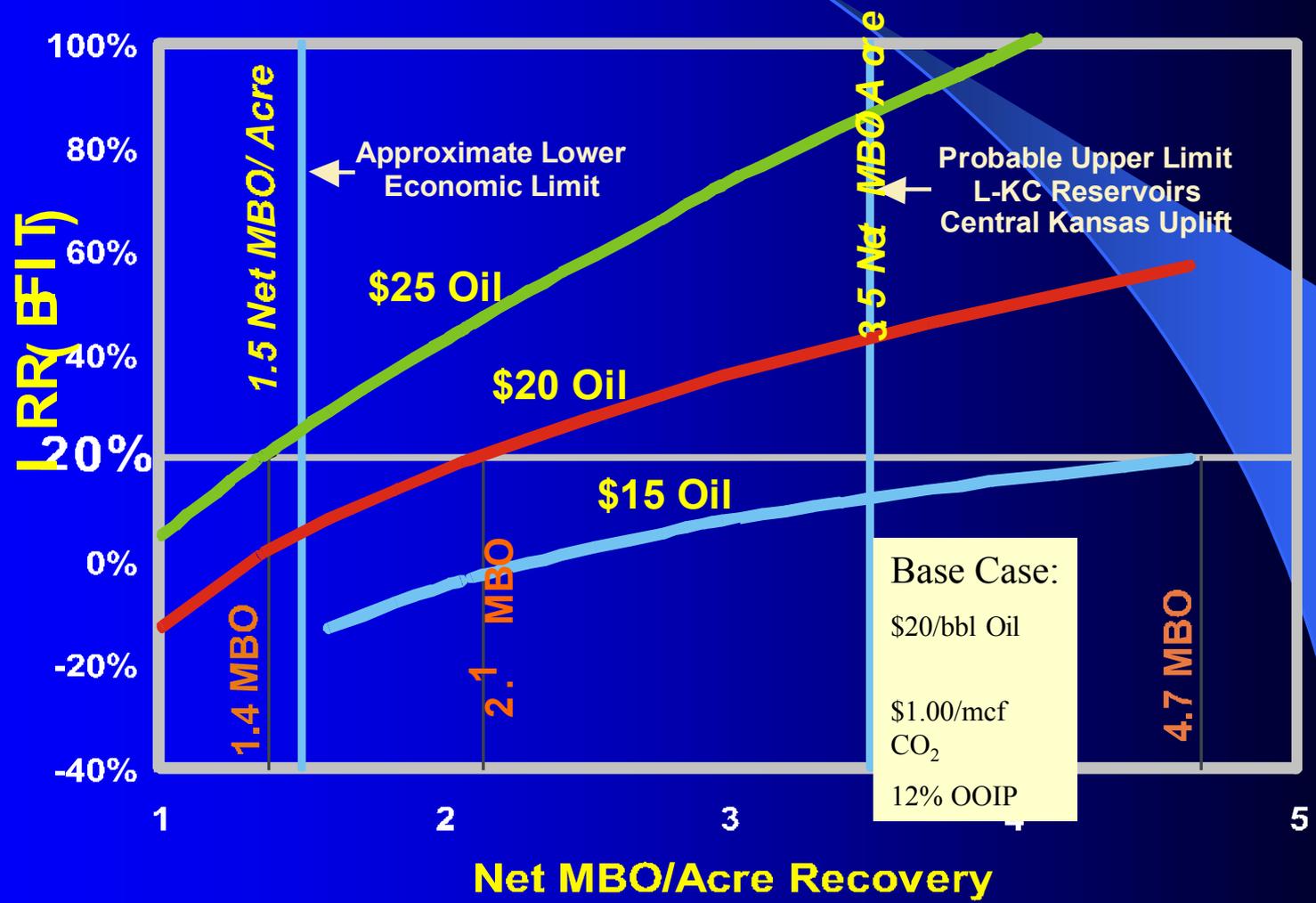
Scripts



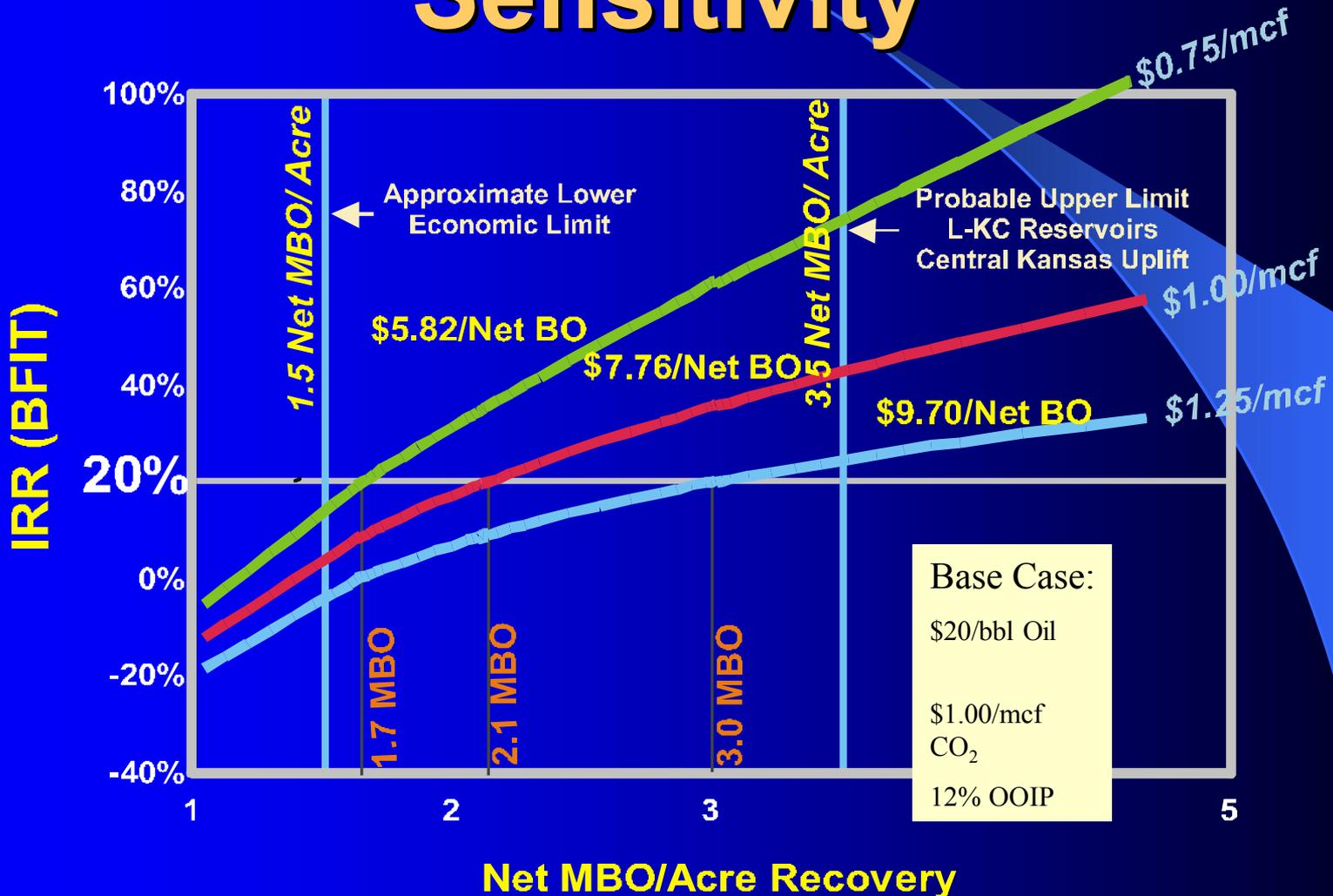
Attributes of Clnn_co2_sps83.shp

Area	Field Name	Thickness	Porosity	CO2
2850301313.737	CANTON CONS.	15.00000	7.00000	51626082545.061
4285518.030	CANTON CONS.	15.00000	7.00000	77621445.318
194849825.051	CANTON CONS.	15.00000	7.00000	3529217456.236
6205116.892	CANTON CONS.	15.00000	7.00000	112390179.706
2606997.778		0.00000	0.00000	0.000
5864018.234	CANTON CONS.	15.00000	7.00000	106212030.263
506368349.835	CANTON CONS.	15.00000	7.00000	9171596736.386
5626921.239	CANTON CONS.	15.00000	7.00000	101917610.941
635351.338	CANTON CONS.	15.00000	7.00000	114749051.110
4620734.257		0.00000	0.00000	0.000
3444833782.952	CANTON CONS.	15.00000	7.00000	62394551893.718

CO₂ Flooding: Oil Price Sensitivity



CO₂ Flooding: CO₂ Price Sensitivity



1995 Carbon Dioxide Emissions for Coal-Fired Utility Units

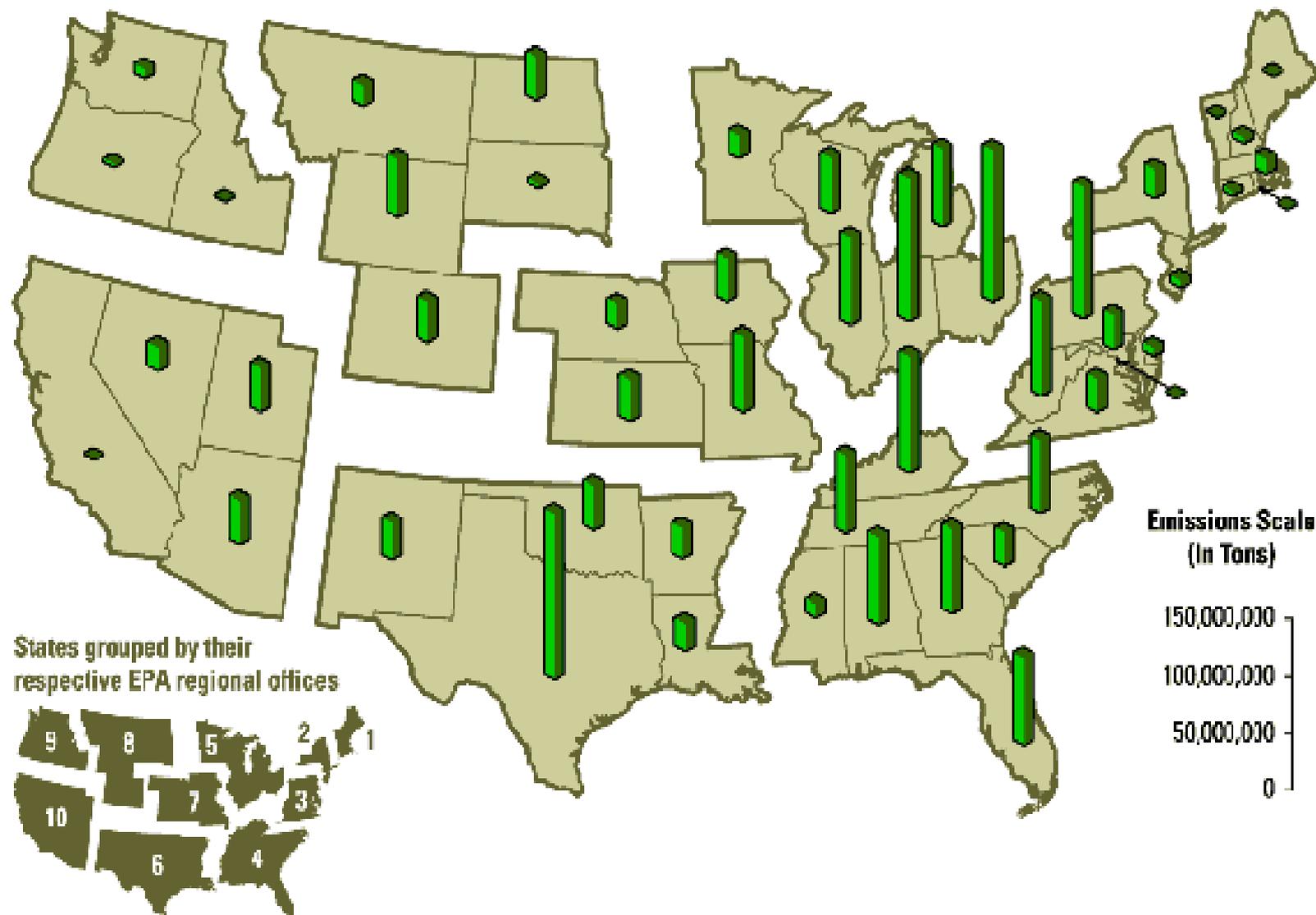


Table 1. Atmospheric CO₂ Data in U.S. Tons

(Source : Pollution Equipment News, June, 2001)

CO ₂ in the earth's atmosphere	5.7 X 10 ¹³ tons/yr
CO ₂ emitted by global soil and vegetarian	4.7 x 10 ¹¹ tons/yr
CO ₂ emitted by the world's oceans	3.6 X 10 ¹¹ tons/yr
CO ₂ emitted globally from fossil fuels	3.2 X 10 ¹⁰ tons/yr
CO ₂ emitted by world's fossil fuel power plants	7.2 X 10 ⁹ tons/yr
CO ₂ emitted by global transportation	5.6 X 10 ⁹ tons/yr
CO ₂ emitted by American power plants	2.3 X 10 ⁹ tons/yr
CO ₂ emitted by the world population breathing	3.3 X 10 ⁹ tons/yr